



PREPARATION OF GRAPHENE/MOLYBDENUM DISULFIDE BASED ELECTRODES AND ITS ELECTROCHEMICAL PERFORMANCE IN SUPERCAPACITORS

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Abstract: Supercapacitor is highly promising energy device due to its electrical charge storage performance and significant lifecycle ability. Construction of the supercapacitor cell especially its electrode fabrication is critical to ensure great application performance. The purpose of this research is to fabricate the molybdenum disulfide (MoS₂), graphene and G/MoS₂ hybrid electrode and their usage as symmetric and asymmetric supercapacitors. The electrode was prepared by using a simple and facile slurry technique. By this, XRD was used to analyze the crystal phase and structure of the as-prepared graphene, MoS₂, and G/MoS₂ hybrid. The peaks at 14.3°, 33.8°, and 57.5° are attributed to the (002), (100), and (110) plane of MoS₂ crystal. From Raman spectroscopy shows the characteristic peaks of graphene (D, G and 2D) and MoS₂ (E_{12g} band at 377 cm⁻¹ and A_{1g} band at 403 cm⁻¹) are retained in the Raman spectra of G/MoS₂ which can confirm the fact that the hybrid of G/MoS₂ is composed of MoS₂ and graphene. Next, the XPS analysis was carried out to deduce the exact elemental composition of the G/MoS₂. The full scan of the G/MoS₂ gives the characteristic peaks for Mo 3d, S 2p, C 1s and O 1s with their corresponding binding energies.



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Publications :

- 1 Formation of Cr₂₃C₆ during the Sensitization of AISI 304 Stainless Steel and its Effect to Pitting Corrosion in International Journal of Electrochemical Science (IJES) 7, 9465-9477.
- 2 Development of high performance electrochemical capacitor: A systematic review of electrode fabrication technique based on different carbon materials in ECS Journal of Solid State Science and Technology 2 (10), M3101.
- 3 Aligned carbon nanotube from catalytic chemical vapor deposition technique for energy storage device: a review in Ionics 19 (11), 1455-1476.
- 4 Significant capacitance performance of vertically aligned single-walled carbon nanotube supercapacitor by varying potassium hydroxide concentration in Int. J. Electrochem. Sci 8 (3), 3902-3911.



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